

# Course Specifications

From the academic year 2019-2020 up to and including the academic year

# Chemical Analytical Methods (0000136)

Due to Covid 19, the education and assessment methods may vary from the information displayed in the schedules and course details. Any changes will be communicated on Ufora.

Course size	(nominal values; actual values may depend on programme)					
Credits 4.0	Study time 120 h	(	Contact hrs	60.0h		
Course offerings and	teaching methods in academic year 2	2021-2022				
A (semester 1)	English	Incheon		lecture		24.0h
				practicum		24.0h
				guided self-study		12.0h
Lecturers in academic	: year 2021-2022					
Cirkovic Velickovic, Tanja			KR01	lecturer-in-charge		
Van Haute, Sam			KR01	co-lecturer		
Offered in the following programmes in 2021-2022			crdts	offering		
Bachelor of Scie	nce in Environmental Technology			4	А	
Bachelor of Scie	nce in Food Technology			4	А	
Bachelor of Scie	nce in Molecular Biotechnology			4	А	
Joint Section Bachelor of Science in Environmental Technology, Food Technology and Molecular Biotechnology				4	А	

# Teaching languages

English

# Keywords

Analytical chemistry, Sampling, Gravimetry, Volumetry, Electrochemical methods, Spectroscopy, Separation techniques, Chromatography

## Position of the course

This course teaches the principles and applications of analytical methods and techniques in the field of bioscience engineering. The transfer of knowledge and the efficient use of these techniques in order to be able to solve analytical problems are the main objectives of the course.

#### Contents

- 1. Introduction
- 2. Basic tools
- 3. Evaluating analytical data
- 4. Standardizing analytical methods
- 5. Equilibrium chemistry
- 6. Collecting and preparing samples
- 7. Gravimetric methods
- 8. Titrimetric methods
- 9. Spectroscopic methods
- 10. Electrochemical Methods

11. Chromatography

Practical sessions: volumetric and gravimetric analysis; potentiometric titration; chromatography.

# Initial competences

Basic knowledge of general and inorganic chemistry.

#### **Final competences**

1 The student will acquire a fundamental knowledge of definitions, units and terminology to describe quantitative and qualitative chemical analytical data, to

have knowledge of the fundamental principles, underlying mechanisms and application potentials of common analytical techniques, to be capable of making an argued choice between possible approaches for addressing an analytical problem and to be able to interpret and report analytical results in an objective way and to evaluate their accuracy and reliability.

- 2 Knowledge of the fundamental principles, underling mechanisms and appliction potentials of common analytical techniques
- 3 Student is able to interpret and report analytical results in an objective way

#### Conditions for credit contract

Access to this course unit via a credit contract is unrestricted: the student takes into consideration the conditions mentioned in 'Starting Competences'

#### Conditions for exam contract

This course unit cannot be taken via an exam contract

# Teaching methods

Practicum, Guided self-study, Lecture

#### Extra information on the teaching methods

Lecture, 24 hours Practicum, 24 hours Guided self-study - problem solving sessions, 12 hours

#### Learning materials and price

Analytical Chemistry 2.0, David Harvey

#### References

#### Course content-related study coaching

Interactive counselling through Minerva. Individual electronic appointment booking.

#### Assessment moments

end-of-term and continuous assessment

## Examination methods in case of periodic assessment during the first examination period

Written examination with multiple choice questions, Written examination with open questions

#### Examination methods in case of periodic assessment during the second examination period

Written examination with multiple choice questions, Written examination with open questions

# Examination methods in case of permanent assessment

Report, Participation, Written examination with multiple choice questions

#### Possibilities of retake in case of permanent assessment

examination during the second examination period is possible in modified form

#### Extra information on the examination methods

Following practical session a written report will be submitted for evaluation and it will contribute to the final mark with 10%. Periodic evaluation (twice during the semester) in a form of a written performance assessment test with open questions will contribute to the mark with 10%. Absence from practical session will result in a maximum mark 9/20.

#### Calculation of the examination mark

Written examination with open questions 60% Written examination with multiple choice questions 20% Performance assessment 10% Report 10%